

Atmospheric Moisture Equations

Symbols List:

L_v	=	$2.50 \times 10^6 \text{ J} \cdot \text{kg}^{-1}$	for water (vaporization)
L_d	=	$2.83 \times 10^6 \text{ J} \cdot \text{kg}^{-1}$	for ice (sublimation)
\mathcal{R}_v	=	$461 \text{ J} \cdot \text{K}^{-1} \cdot \text{kg}^{-1}$	(gas constant for water vapor)
\mathcal{R}_d	=	$287 \text{ J} \cdot \text{K}^{-1} \cdot \text{kg}^{-1}$	(gas constant for dry air)
ϵ	=	$\mathcal{R}_d / \mathcal{R}_v = 0.622$	g _{vapor} / g _{dry air}
ρ_d	=	density of dry air at sea level and 20°C	$= 1.22 \text{ kg} \cdot \text{m}^{-3}$
P	=	total atmospheric pressure	
T	=	parcel temperature	
e	=	vapor partial pressure	
T_0	=	273 K	
e_0	=	0.611 kPa = 6.11 mb or hPa	

◇ **Saturation Vapor Pressure** e_s

Clausius-Clapeyron equation for atmospheric conditions

$$\frac{de_s}{dT} = \frac{L_v(T)e_s}{R_v T^2}$$

Solving the differential equation results in the following expression for saturation vapor pressure e_s as a function of temperature T :

$$e_s = e_0 \exp \left[\frac{L}{\mathcal{R}_v} \left(\frac{T - T_0}{T_0 \cdot T} \right) \right] \quad \underset{=}{=} \quad \text{units of } e_0, \quad L = L_v \text{ for water surfaces, and } L_d \text{ for ice surfaces}$$

◇ **Mixing Ratio** MR

$$MR = \frac{\text{mass of water vapor}}{\text{mass of dry air}} = \frac{\epsilon \cdot e}{P - e} \quad \underset{=}{=} \quad \frac{\text{g of vapor}}{\text{g of dry air}}$$

◇ **Specific Humidity** SH

$$SH = \frac{\text{mass of water vapor}}{\text{total mass of air}} = \frac{\epsilon \cdot e}{P} \quad \underset{=}{=} \quad \frac{\text{g of vapor}}{\text{g of air}}$$

◇ **Absolute Humidity** AH

$$AH = \frac{\text{mass of water vapor}}{\text{volume of air}} = \frac{e}{\mathcal{R}_v T} = \frac{e}{P} \epsilon \rho_d \quad \underset{=}{=} \quad \frac{\text{kg of vapor}}{\text{m}^3}$$

◇ **Relative Humidity** RH

$$RH = \frac{e}{e_s} \times 100\% = \frac{MR}{MR_s} \times 100\% = \frac{SH}{SH_s} \times 100\% = \frac{AH}{AH_s} \times 100\%$$

$()_s$ = saturation value for the given quantity

◇ **Virtual Temperature** T_v

$$T_v = T(1 + 0.61MR^*)$$

* - the mixing ratio (MR or W) must be in kg/kg, not g/kg